

# CASE

*Curriculum for Agricultural  
Science Education*

**Principles of Agricultural Science – Plant**

# Principles of Agricultural Science – Plant

## A Plant's Grocery Store

Unit 6 – The Growing Environment  
Lesson 6.1 Plant Food

# Sources of Plant Nutrients



- Organic substances
- Legumes
- Chemical fertilizers

# Organic Substances



- Examples: manure, sludge, and compost
- Manure contains nutrients in varying amounts.
- Sludge is **processed** human waste.
- Composting raises organic matter content and destroys pathogens.

# Legumes



- Examples: peanuts, soybeans, and vetch
- Legumes “fix” nitrogen.
- Improve nitrogen content of the soil.
- Most other crops deplete nitrogen.
- Often rotated with nitrogen-depleting crops.

# Chemical Fertilizers

Manufactured fertilizers

Inorganic compounds concentrate desired nutrients

<b>Benefits</b>	<b>Disadvantages</b>
<ul style="list-style-type: none"><li>• Easy to apply</li><li>• Concentrated</li><li>• Readily available</li><li>• Formulated to plant needs</li></ul>	<ul style="list-style-type: none"><li>• Expensive</li><li>• Petroleum based</li></ul>

# Forms of Chemical Fertilizers



Commercial chemical fertilizer comes in different forms:

- Dry (granular or pelleted)
- Liquid
- Water soluble powder

# What the Numbers Mean



The set of three numbers of a fertilizer analysis is the N-P-K percentage.

Example: A bag of 10-20-10 means the bag contains:

- 10% elemental nitrogen
- 20% phosphate ( $P_2O_5$ )
- 10% potash ( $K_2O$ )

# Let's do some math...



How much nitrogen is in a 100 lb. bag of 15-16-17 fertilizer?

Correct: 15 lbs.

$$15\% \times 100 = .15 \times 100 = 15 \text{ lbs.}$$

# Calculating Fertilizer Application Rates

How many pounds of fertilizer do you need to apply in order to get enough nutrients for a plant?

$$\frac{\text{Value of N-P-K from Fertilizer}}{100 \text{ Pounds}} = \frac{\text{Nutrient Rate}}{\text{X lbs of Fertilizer}}$$

# Let's try a rate calculation...



You want to add fertilizer to a family member's yard.

- They have 1000 square feet of lawn
- Recommended amount of nitrogen application is 1 lb /1000 sq. ft.
- The fertilizer you have is 16-16-16

How much **nitrogen** fertilizer should you apply?

# Problem Solved

Set up the formula:

$$\frac{16 \text{ lbs. of N}}{100 \text{ Pounds}} = \frac{1 \text{ lb. N/1000 sqft.}}{X \text{ lbs of Fertilizer}}$$

$$16x = 100$$

**x = 6.25 lbs. of fertilizer  
for every 1000 sq. ft.**

# Calculations Depends on Form



- Dry fertilizers are expressed in weight measurements
- Liquid or water soluble fertilizers are expressed in parts per million (ppm)

# Fertilizer Cost



Cost can be confusing...

Why does a 100 lb bag of 20-20-20 cost twice as much as a 100 lb bag of 10-10-10?

Nutrient content is based on percentage, 20-20-20 has twice as much nutrient value as 10-10-10.

# References



Parker, R. (2010). *Plant and soil science: Fundamentals and applications*. Clifton Park, NY: Delmar.

Plaster, Edward J. (1992). *Soil science and management (2<sup>nd</sup> Ed.)*. Albany, NY: Delmar.