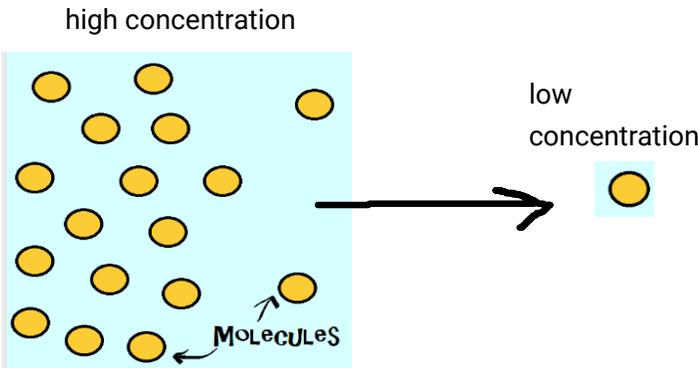


### Amoeba Sisters Video Recap of *Osmosis*

1. The below picture represents **diffusion** of molecules. Place the following labels in the diagram: **high concentration**, **low concentration**, and **an arrow** showing the direction that the molecules would travel before equilibrium is reached.



2. **Osmosis** is a type of diffusion, but it involves the movement of water. Similar to diffusion, osmosis is the movement of molecules (water molecules if osmosis) from a high concentration to a low concentration.

The video clip explains that you can also look at water as moving to a higher concentration of **solute** molecules.

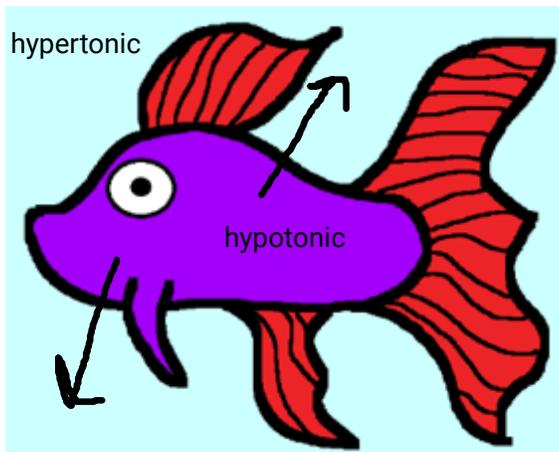
Why can it also be viewed this way?

It can be viewed this way because if one area has a higher concentration of solute molecules it has less molecules of water so water will move that way to balance out and find an equilibrium.

3. **Osmosis Scenario:** The video clip mentioned a disaster scenario of a saltwater fish being placed in fresh water.

What would occur if, instead, a freshwater fish was placed in saltwater? The water in the fish would leave to go towards the saltwater and the fish would dehydrate and eventually die.

Your answer needs to have an **arrow** indicating the direction of water flow in osmosis, a label for **"hypertonic,"** and a label for **"hypotonic."**



4. **Osmosis Scenario:** Fluid movement into the brain after traumatic brain injury can result in dangerous brain swelling.

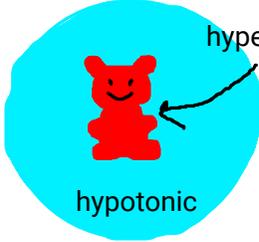
One treatment that can be used in some of these cases is adding a hypertonic saline. You need to decide whether this blank should be the word hypertonic or hypotonic. Remember, you are trying to reduce the excessive fluid in the brain.

Explain your answer:

If we want to prevent water from flooding into the brain and causing swelling we should put a hypertonic solution into something like an IV to make the rest of the body have a higher concentration which would pull the water away from the brain and decrease swelling.

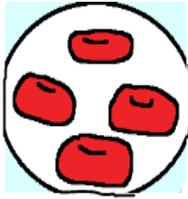
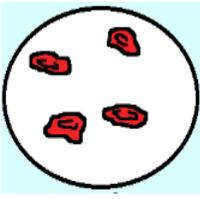
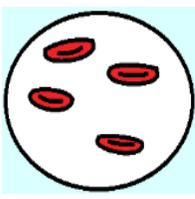


### The Gummy Bear Mystery

<p>Do you like gummy bears? We do! They are one of our favorite snacks, though we (try to) eat them in moderation because they are high in sugar. Consider that your sister is in a foul mood and decides to dump your gummy bears in your ice water about 30 minutes before you get home.</p> <p>The gummy bears are greatly enlarged by the time you get home! Your sister and some friends have different viewpoints for what happened. Draw a diagram below showing the enlarged gummy bears in a cup of water. Place the labels "hypertonic" and "hypotonic" in your diagram. One label should be for the gummy bears and one label should be for the water.</p> <p>5. </p>	<b>Viewpoints:</b>	<p>6. Whose viewpoint is correct in the viewpoint column? A good answer has a good defense! <b>Defend your answer</b>, and also <b>give reasons why the other explanations are incorrect.</b></p> <p>My friend Joe is correct because the gummy bears had a higher concentration than the water because of the sugar causing the water to move from its low concentration to the higher concentration in the gummy bears making them become enlarged. The other answers are incorrect because if the gummy bears were hypotonic then the water would've left the gummy bears causing them to shrink.</p>
	A) Your sister said that the sugar left the gummy bears, because the gummy bears were <b>hypertonic</b> compared to the water.	
	B) Your friend Joe said that water traveled into the gummy bears, because the gummy bears were <b>hypertonic</b> compared to the water.	
	C) Your friend Suzy said the sugar went into the gummy bears, because the gummy bears were <b>hypotonic</b> compared to the water.	
	D) Your friend Will said that water traveled into the gummy bears by osmosis, because the gummy bears were <b>hypotonic</b> compared to the water.	

### Hypertonic, Hypotonic, or Isotonic? Oh My!

These red blood cells have all been placed in different solutions! Based on their appearance after being placed in these solutions for a period of time, place on each line (A) for **hypertonic**, (B) for **hypotonic**, or (C) for **isotonic**.

<p>7. The cells are <u>hypertonic</u> compared to the <u>hypotonic</u> solution.</p>  <p style="text-align: right;"><b>SWELLING</b></p>	<p>8. The cells are <u>hypotonic</u> compared to the <u>hypertonic</u> solution.</p>  <p style="text-align: right;"><b>SHRINKING</b></p>	<p>9. The cells are <u>isotonic</u> compared to the <u>isotonic</u> solution.</p>  <p style="text-align: right;"><b>STABLE</b></p>
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