

Sergio Neto

Step 1- I have made a Atom.

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0 ? 0

positive ion

Li Build

An interactive whiteboard activity where students build an atom by dragging electrons, neutrons and protons onto the template. The element information box shows if they are correct. The animation can also be used to show how ions form. Enter symbols in the build text box to produce any of the first 20 elements.

Step 2 I have created a Lithium (Li)

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7
Li
3

Lithium

neutral atom

Li Build

An interactive whiteboard activity where students build an atom by dragging electrons, neutrons and protons onto the template. The element information box shows if they are correct. The animation can also be used to show how ions form. Enter symbols in the build text box to produce any of the first 20 elements.

Step 3- I have made a Carbon (C) atom

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12 ⁶⁺
C
6
Carbon
positive ion
Li Build

An interactive whiteboard activity where students build an atom by dragging electrons, neutrons and protons onto the template. The element information box shows if they are correct. The animation can also be used to show how ions form. Enter symbols in the build text box to produce any of the first 20 elements.

Step 4- Both neutrons and electrons changes the type of elements as you add it to the atom.

Step 5- I have created a Hydrogen atom.

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An interactive whiteboard activity where students build an atom by dragging electrons, neutrons and protons onto the template. The element information box shows if they are correct. The animation can also be used to show how ions form. Enter symbols in the build text box to produce any of the first 20 elements.

I have created a Helium atom.

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An interactive whiteboard activity where students build an atom by dragging electrons, neutrons and protons onto the template. The element information box shows if they are correct. The animation can also be used to show how ions form. Enter symbols in the build text box to produce any of the first 20 elements.

Step 6- What makes an atom stable is the number of electrons on the outer shell determines how reactive.

What makes an atom unstable is the unbalanced forces, if the nucleus has an excess of internal energy.

What makes an atom become ion is an atom that have lost or gained an electrons.