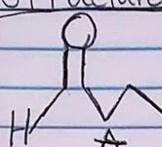
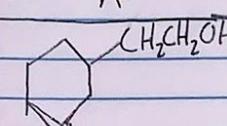
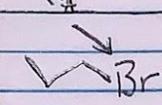
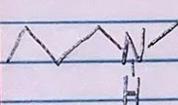
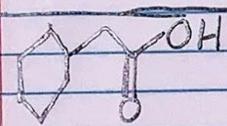
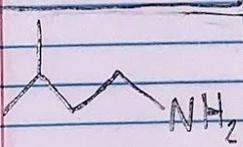
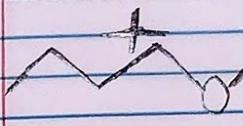
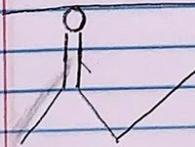


# Question 1

## Topic 4 Homework Question #1

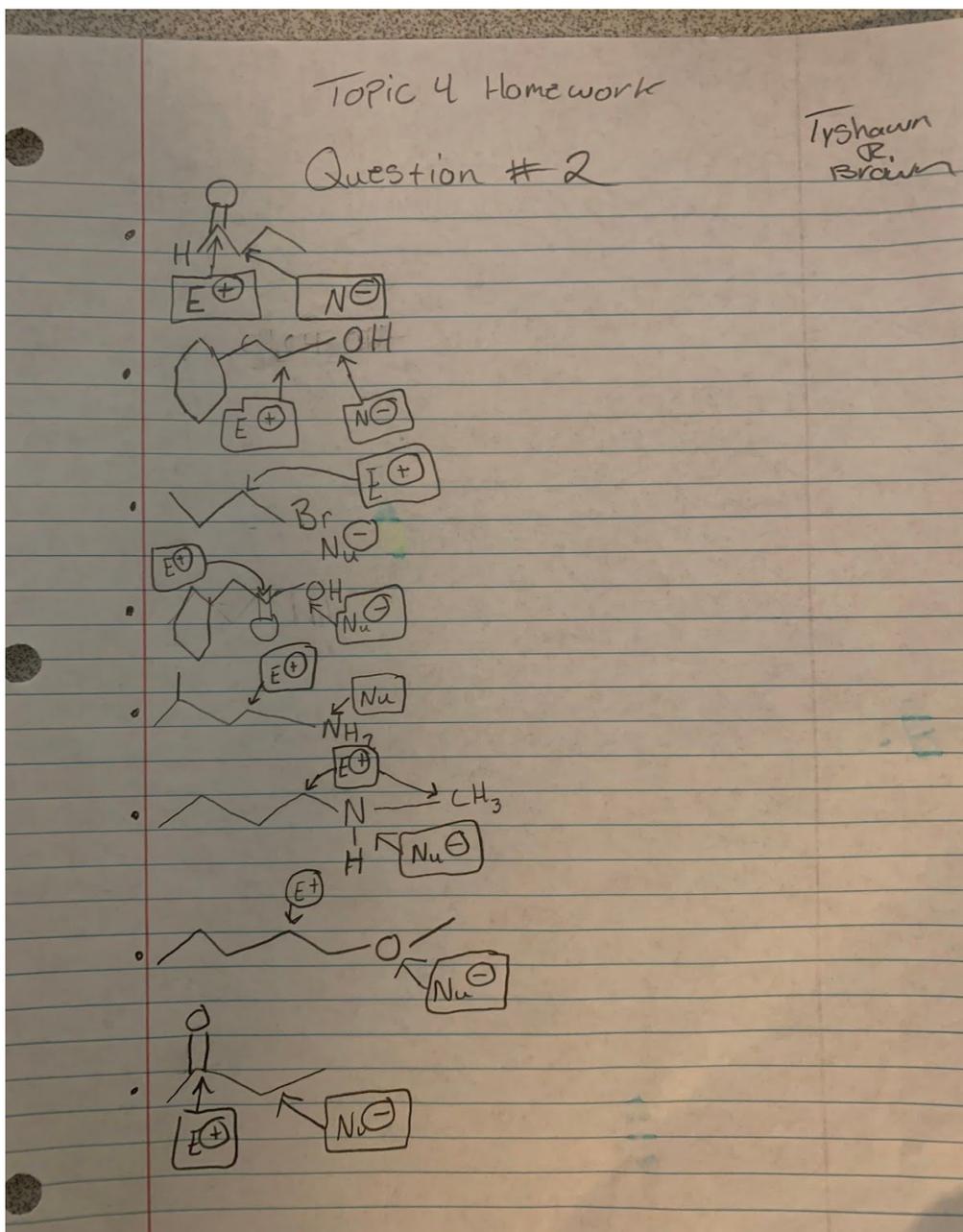
Tyshawn  
R.  
Brown

| Structure  | Functional group | Compound name                       |
|--|------------------|-------------------------------------|
|               | Aldehyde         | Butan-1-al                          |
|               | Alcohol          | 1-Cyclohexylethanol                 |
|               | Alkyl bromide    | 1-Bromopropane                      |
|              | 2° Amine         | N-Methylbutan-2-amine               |
| <p>(A)</p>  | Carboxylic Acid  | 1-Cyclohexylethanoic acid           |
| <p>(B)</p>  | 1° Amine         | 3-methylbutan-1-amine               |
| <p>(C)</p>  | Ether            | Butyl methyl ether or methoxybutane |
| <p>(D)</p>  | Ketone           | 2-butanone                          |

## Question 2

Electrophilic site = E +

Nucleophilic site = N -



# Questions # 3 - 4

Topic 4

# 3-4 Question

# 3 Intermolecular Force

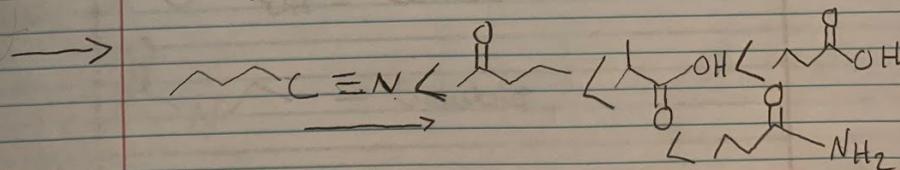
- (A)  $\text{CO}_2$  - London dispersion Forces
- (B)  $\text{H}_2\text{C} - \text{CH}_2 - \text{CH}_3$  - Vander Waals forces
- ✓ (C)  $\text{NaF}$  - Electrostatic Forces / H-bond
- (D)  $\text{H}_3\text{C} - \text{Cl}$  - Dipole - Dipole interaction

Stronger

- ✓ (E)  $\text{H}_3\text{C} - \text{CH}_2 - \text{OH}$  - H-bonding (intermolecular)

- (F)  - Vander waals, dipole - dipole

#4 Boiling Point order increasing order



# Question # 5

# 5 Question

Topic 4 Homework

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#5

Water Solubility

